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Covalent Bond - an overview | ScienceDirect Topics

(Source: McMurry J and Fay RC (2003) Chemistry, 4th edition Prentice Hall, Upper Saddle River, NJ. Reprinted with the permission of Prentice Hall, Upper Saddle River, New Jersey, USA.) In a single covalent bond, when the electrons are shared between two s orbitals, the resulting bond is a sigma (σ) bond as shown in Figure 3-4 .

Nomenclature of Cycloalkanes | MCC Organic Chemistry

The most common and useful cycloalkanes in organic chemistry are cyclopentane and cyclohexane, although other cycloalkanes varying in the number of carbons can be synthesized. Understanding cycloalkanes and their properties are crucial in that many of the biological processes that occur in most living things have cycloalkane-like structures.

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Nomenclature of Coordination Complexes - Chemistry LibreTexts

Rule 3: Ligand Multiplicity. The number of ligands present in the complex is indicated with the prefixes di, tri, etc. The exceptions are polydentates that have a prefix already in their name (en and EDTA 4- are the most common). When indicating how many of these are present in a coordination complex, put the ligand's name in parentheses and use bis (for two ligands), tris (for three ligands) ...

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